



تفاصيل البحث:

Analysis of domperidone in pharmaceutical formulations and wastewater by differential pulse voltammetry at a glassy-carbon electrode

Analysis of domperidone in pharmaceutical formulations and wastewater by differential pulse voltammetry at a glassy-carbon electrode

The redox characteristics of the drug domperidone at a glassy-carbon electrode (GCE) in aqueous media were critically investigated by differential-pulse voltammetry (DPV) and cyclic voltammetry (CV). In Britton Robinson (BR) buffer of pH 2.6-10.3, an irreversible and diffusion-controlled oxidation wave was developed. The dependence of the CV response of the developed anodic peak on the sweep rate (v) and on depolizer concentration was typical of an electrode-coupled chemical reaction mechanism (EC) in which an irreversible first-order reaction is interposed between the charges. The values of the electron-transfer coefficient (α) involved in the rate-determining step calculated from the linear plots of $E_p(a)$ against $\ln(v)$ in the pH range investigated were in the range 0.64 ± 0.05 confirming the irreversible nature of the oxidation peak. In BR buffer of pH 7.6-8.4 a well defined oxidation wave was developed and the plot of peak current height of the DPV against domperidone concentration at this peak potential was linear in the range 2.40×10^{-6} to 2.40×10^{-5} mol L⁻¹ with lower limits of 5.20×10^{-7} mol L⁻¹ for detection (LOD) and 6.1×10^{-7} mol L⁻¹ for quantitation (LOQ), respectively. A relative standard deviation of 2.39% ($n = 5$) was obtained for 8.5×10^{-6} mol L⁻¹ of the drug. These DPV procedures were successfully used for analysis of domperidone in the pure form (98.2 \pm 3.1%), dosage form (98.35 \pm 2.9%), and in tap (97.0 \pm 3.6%) and wastewater (95.0 \pm 2.9%) samples. The method was validated by comparison with standard titrimetric and HPLC methods. Acceptable error of less than 3.3% was also achieved.

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